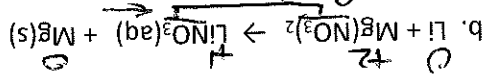
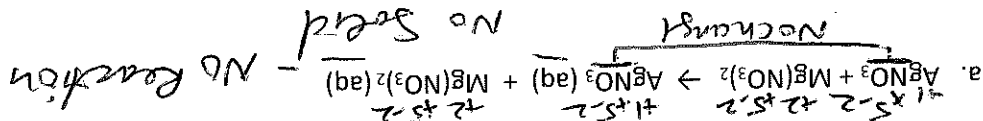
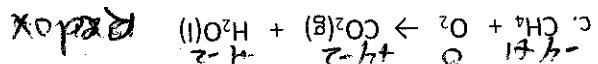


Preliminary Quiz Test Ticket  
Redox Reactions

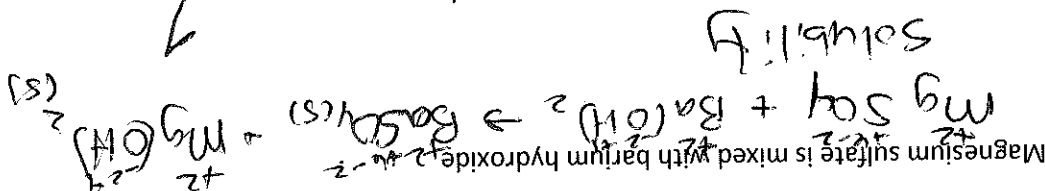
1. In the following reactions determine if the reaction is a Redox/solubility reaction/No reaction.



The NO<sub>3</sub><sup>-</sup> did not change



d. Molecular equation:



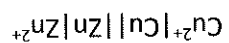
e. Which ions are spectators?

None → Both precipitate

A voltaic cell is using the following half reaction notation to generate a voltage.

a. How much voltage can be produced? **1.10V**

The notation below is different but I think you can figure it out!



b. Who is being oxidized?

Zinc **LEO**

e<sup>-</sup> on product side

