

Solubility product constants @ 25°C			
Compound	Ksp	Compound	Ksp
AgCl	1.6E-10	Fe(OH) ₂	8E-16
AgBr	5E-13	Hg ₂ Br ₂	6E-23
AgC ₂ H ₃ O ₂	1.8E-3	Hg ₂ Cl ₂	1.4E-18
Ag ₂ C ₂ O ₄	5E-12	Hg ₂ SO ₄	7E-7
Ag ₂ CrO ₄	1.12E-12	Hg ₂ CO ₃	1.1E-3
AgI	9E-17	MgF ₂	3.7E-8
AgSCN	1.0E-12	Mn(OH) ₂	1.7E-13
Ag ₂ SO ₄	1.0E-5	Mg(OH) ₂	1.8E-11
BaCrO ₄	8 E-11	NiCO ₃	1.2E-7
BaSO ₄	1.1E-10	PbBr ₂	1.2E-5
CaCO ₃	5 E-9	PbCl ₂	1.5E-5
CaF ₂	1.4E-10	PbCO ₃	1.2E-13
Ca(OH) ₂	5E-6	PbI ₂	8.49 x 10 ⁻⁹
CaSO ₄	7E-5	PbSO ₄	6.3E-7
		SrSO ₄	3E-7
		SrF ₂	4E-9

Acid and base equilibrium constants			
Compound	Chemical formula	Equation	Ka
Inorganic acids			
Chlorous acid	HClO ₂	$\text{HClO}_2 + \text{H}_2\text{O} \rightarrow \text{ClO}_2^- + \text{H}_3\text{O}^+$	1.1E-2
Nitrous acid	HNO ₂	$\text{HNO}_2 + \text{H}_2\text{O} \rightarrow \text{NO}_2^- + \text{H}_3\text{O}^+$	7.2E-4
Hydrofluoric acid	HF	$\text{HF} + \text{H}_2\text{O} \rightarrow \text{F}^- + \text{H}_3\text{O}^+$	6.6E-4
Hypochlorous acid	HOCl	$\text{HOCl} + \text{H}_2\text{O} \rightarrow \text{OCl}^- + \text{H}_3\text{O}^+$	2.9E-8
Hypobromous acid	HBrO	$\text{HBrO} + \text{H}_2\text{O} \rightarrow \text{BrO}^- + \text{H}_3\text{O}^+$	2.5E-9
Hydrocyanic acid	HCN	$\text{HCN} + \text{H}_2\text{O} \rightarrow \text{CN}^- + \text{H}_3\text{O}^+$	6.2E-10
Carboxylic acids			
Chloroacetic acid	CH ₂ ClCOOH	$\text{CH}_2\text{ClCOOH} + \text{H}_2\text{O} \rightarrow \text{CH}_2\text{ClCOO}^- + \text{H}_3\text{O}^+$	1.4E-3
Formic acid	HCOOH	$\text{HCOOH} + \text{H}_2\text{O} \rightarrow \text{HCOO}^- + \text{H}_3\text{O}^+$	1.8E-4
Benzoic acid	C ₆ H ₅ COOH	$\text{C}_6\text{H}_5\text{COOH} + \text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_5\text{COO}^- + \text{H}_3\text{O}^+$	6.3E-5
Acetic acid	CH ₃ COOH	$\text{CH}_3\text{COOH} + \text{H}_2\text{O} \rightarrow \text{CH}_3\text{COO}^- + \text{H}_3\text{O}^+$	9.1E-5
Inorganic Bases			
			Kb
Ammonia	NH ₃	$\text{NH}_3 + \text{H}_2\text{O} \rightarrow \text{NH}_4^+ + \text{OH}^-$	1.8E-5
Hydrazine	H ₂ NNH ₂	$\text{H}_2\text{NNH}_2 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{NNH}_3^+ + \text{OH}^-$	8.5E-7
Hydroxylamine	HONH ₂	$\text{HONH}_2 + \text{H}_2\text{O} \rightarrow \text{HONH}_3^+ + \text{OH}^-$	9.1E-9
Amines			
Dimethylamine	(CH ₃) ₂ NH	$(\text{CH}_3)_2\text{NH} + \text{H}_2\text{O} \rightarrow (\text{CH}_3)_2\text{NH}_2^+ + \text{OH}^-$	5.9E-4
Ethylamine	CH ₃ CH ₂ NH ₂	$\text{CH}_3\text{CH}_2\text{NH}_2 + \text{H}_2\text{O} \rightarrow \text{CH}_3\text{CH}_2\text{NH}_3^+ + \text{OH}^-$	4.3E-4
Methylamine	CH ₃ NH ₂	$\text{CH}_3\text{NH}_2 + \text{H}_2\text{O} \rightarrow \text{CH}_3\text{NH}_3^+ + \text{OH}^-$	4.2E-4
Pyridine	C ₅ H ₅ N	$\text{C}_5\text{H}_5\text{N} + \text{H}_2\text{O} \rightarrow \text{C}_5\text{H}_5\text{N}^+ + \text{OH}^-$	1.5E-9
aniline	C ₆ H ₅ NH ₂	$\text{C}_6\text{H}_5\text{NH}_2 + \text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_5\text{NH}_3^+ + \text{OH}^-$	7.4E-10