

# Section Review

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## Behavior of Gases

### USING KEY TERMS

1. Use each of the following terms in the same sentence: *temperature*, *pressure*, *volume*, and *Charles's law*.

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### UNDERSTANDING KEY IDEAS

- \_\_\_\_\_ 2. Boyle's law describes the relationship between
- a. volume and pressure.
  - b. temperature and pressure.
  - c. temperature and volume.
  - d. All of the above

3. What are the effects of a warm temperature on gas particles?

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### MATH SKILLS

- 4 You have 3 L of gas at a certain temperature and pressure. What would the volume of the gas be if the temperature doubled and the pressure stayed the same? Show your work below.

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**CRITICAL THINKING**

5. **Applying Concepts** What happens to the volume of a balloon that is taken outside on a cold winter day? Explain.

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6. **Making Inferences** When scientists record a gas's volume, they also record its temperature and pressure. Why?

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7. **Analyzing Ideas** What happens to the pressure of a gas if the volume of gas is tripled at a constant temperature?

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