NameDate		
Directed Reading A		
Section: Compounds 1. List three examples of compounds you encounter every day.		
COMPOUNDS: MADE OF ELEMENTS		
 2. Which of the following is NOT true about compounds? a. Compounds are combinations of elements that join in specific ratios according to their masses. b. The mass ratio of a specific compound is always the same. c. Compounds are random combinations of elements. d. Different mass ratios mean different compounds. 		
3. When two or more elements are joined by chemical bonds to form a new pure		
substance, we call that new substance a(n)		
4. A compound is different from the that reacted to		
form it.		
PROPERTIES OF COMPOUNDS		
 5. Which of the following statements is true about the properties of compounds? a. A property of all compounds is to react with acid. b. Each compound has its own physical properties. c. Compounds cannot be identified by their chemical properties. d. A compound has the same properties as the elements that form it. 		
6. Sodium and chlorine can be extremely dangerous in their elemental form. How is it possible that we can eat them in a compound?		

Name	Class	Date
Directed Reading A cont	inued	
Match the correct descript provided.	ion with the correct tern	n. Write the letter in the space
7. a poisonous, gro	eenish yellow gas	a. sodium chloride
8. table salt		b. chlorine
9. a soft, silvery w violently with v		c. sodium
BREAKING DOWN COM	POUNDS	
10. What compound helps	give carbonated bevera	ges their "fizz"?
11. Which elements make their "fizz"?	up the compound that h	elps give carbonated beverages
10 Til 1 1	1 1 1	
12. The only way to break		rougn
a(n)	change.	
COMPOUNDS IN YOUR	WORLD	
13. Aluminum is produced		compound
		in photosynthesis to make
carbohydrates.		