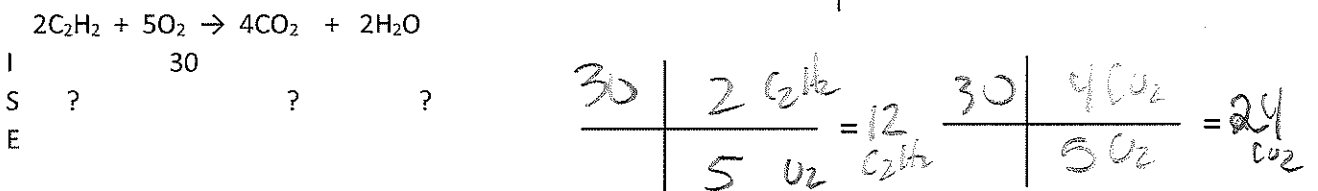
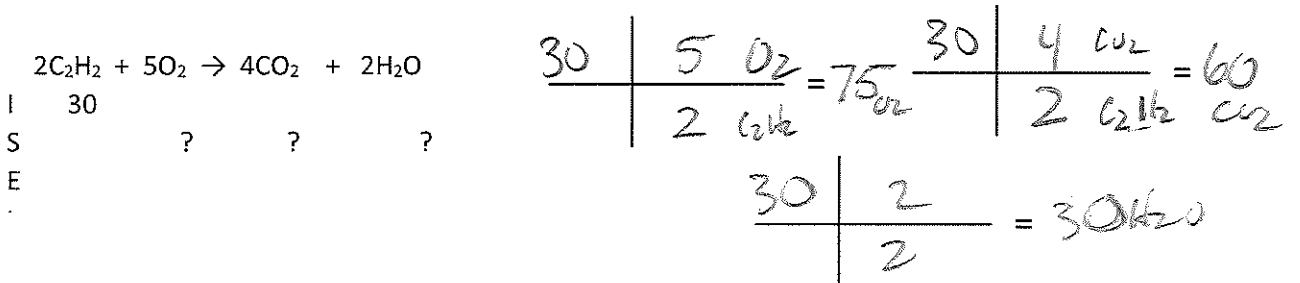
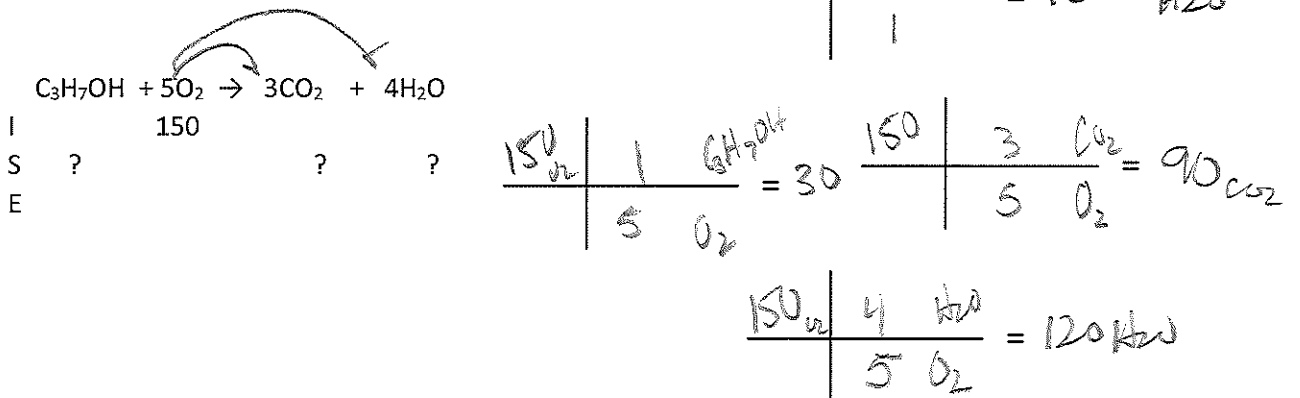
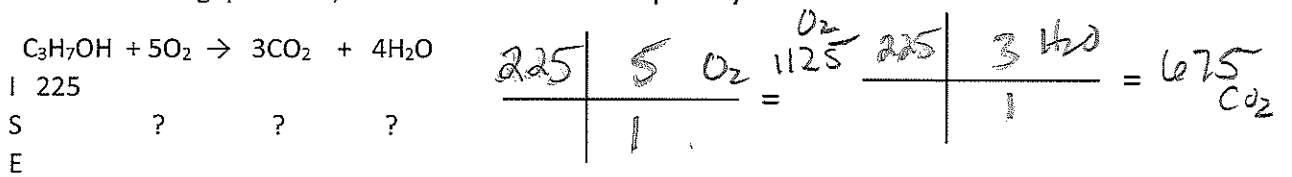


Stoichiometric Ratio Particles 1

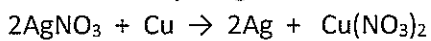
Objective: Be able to set up an ISE table and utilize the stoichiometric ratio.

For each of the following questions, calculate the ? unknown quantity.

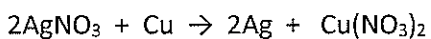


For the following 2, you are given 80 of a single reactant.

Solve for everything else like above.



$$\frac{80}{2} \bigg| \frac{2}{2} = 80 Ag$$



$$\frac{80}{2} \bigg| \frac{1}{2} = 40 Ag$$

$$\frac{30}{5} \bigg| \frac{4 H_2O}{5} = 24 H_2O$$